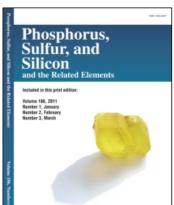
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Synthesis and Conducting Properties of TetracyanoazulenequinodimethaneTetrathiotetracene Complexes

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Synthesis and Conducting Properties of Tetracyanoazulenequinodimethane-Tetrathiotetracene Complexes

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Tetracyanoquinodimethane (TCNQ) and their derivatives have proved to be the most impotant π -acceptor molecules for organic conductors. We are interested in new nonbenzenoid TCNQ-type acceptors, and now synthesized tetracyanoazulenequinodimethanes (TCNAzQDMs: **1a,b** and **2a,b**), and have clarified conducting properties of their tetrathiotetracene (TTT) complexes, which are reported herein. Synthesis of the new acceptors was carried out by the reaction of 1,3,5-tribromo-azulene (Br₃-Az) with tetracyanoethylene oxide (TCNEO), affording

SCHEME 1

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TABLE I Appearances, Conductivities, Activation Energy, and IR Data of TCNAzQDM-TTT Complexes

Complex	Appearance	$\sigma_{r.t.} S \cdot cm^{-1}$	$E_{ m a}/{ m eV}$	$\nu_{\rm CN}/{\rm cm}^{-1}$
[1a][TTT] ₂ [1b][TTT] ₂ [TCB] _{0.5}	Black powder Black powder	$\begin{array}{c} 3.0 \times 10^{-6} \\ 5.3 \times 10^{-2} \end{array}$	9.9×10^{-2}	2205 2208
[2b] [TTT]	Black violet fine needles	$3.8 imes 10^{-3}$		2196

3-bromo-TCN-1,5-AzQDM **1a**, -1,7-AzQDM **1b** and their dibromo derivatives **2a,b** in 14, 14, 4, and 4% yields, respectively.

Preparation of charge-transfer complexes was performed by the direct reactions of $\mathbf{1a}$ and $\mathbf{1b}$ with TTT in 1,2,4-trichlorobenzene (TCB). The obtained $[\mathbf{1b}][\mathrm{TTT}]_2[\mathrm{TCB}]_{0.5}$ showed a room-temperature conductivity of $5.3 \times 10^{-2}~\mathrm{S}\cdot\mathrm{cm}^{-1}$, which is 10^4 times higher than that of $[\mathbf{1a}][\mathrm{TTT}]_2$.

The reaction of **2b** with TTT in 1,2,4-TCB afforded [**2b**][TTT] complex as black violet fine needles. From the X-ray analysis, mixed-stacks of **2b** and TTT molecules were revealed.